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### chapter 12 stoichiometry test pdf

Chapter 12 Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. \_\_\_\_ 1. The first step in most stoichiometry problems is to \_\_\_\_.

a. add the coefficients of the reagents  
c. convert given quantities to volumes  
b. convert given quantities to moles  
d. convert given quantities to masses

\_\_\_\_ 2. In the reaction  $2\text{CO}(g) + \text{O}_2(g) \rightarrow 2\text{CO}_2(g)$  ...

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20 Then do some stoichiometry using "easy math" • 16 g of methane (MM = 16) is 1 mole and 1 mole of methane will produce 1 mole of CO<sub>2</sub> = 44 g, and 2 moles of H

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